KEYWORDS P B

<u>Halogens:</u> the elements in Group VII of the Periodic Table.

Halide: Compound containing an ion formed when a halogen atom has gained an electron to complete its outer shell of electrons.



<u>Diatomic:</u> Molecule containing two atoms.

Halogen Displacement Reactions

Displacement: a more reactive halogen will take the place of a less reactive halogen from its compound.