

| Name: | |
|-------|--|
| Date: | |

If copper oxide is heated with magnesium it will react to form magnesium oxide and copper. We can write this as a simple word equation:

| Copper oxide | + Magnesium | Magnesium oxide | + | Copper |
|--------------|-------------|---------------------|---|--------|
| | | | | |

Reactants Products

(substances we started the reaction with), always go on the **left**.

(substances formed during a reaction), always go to the **right** of the arrow.

For each of the chemical reactions below:

- 1. Highlight or circle the reactants in one colour and the products in another colour,
- 2. Write the word equation and balanced chemical equation for each reaction.

| Reaction 1 |
|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Kiara dropped a lump of magnesium into a test tube of hydrochloric acid, the chemical magnesium chloride was formed. She also noticed a gas being formed. When a lit splint was placed in the gas it made a squeaky 'pop' sound. |
| |
| Reaction 2 |
| Lucy reacted a lump of calcium carbonate with some sulphuric acid. Two of the products formed were water and calcium sulphate. The other product formed was a gas that turned lime water milky. |
| |
| Reaction 3 |
| Milos put a brand new iron nail outside and studied it for several weeks. He noticed that gradually the surface of the nail became coated with rust. His teacher has told him that the chemical name for rust is "iron oxide". |
| |



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Use the table of results below to help you work out the answers to the questions.

| | Silver nitrate | Copper sulfate | Magnesium sulfate | Iron sulfate | Zinc sulfate |
|-----------|----------------|-------------------|-------------------|--------------|--------------|
| silver | х | no reaction | no reaction | no reaction | no reaction |
| copper | reaction | Х | no reaction | no reaction | no reaction |
| magnesium | reaction | reaction | х | reaction | reaction |
| iron | reaction | reaction | no reaction | х | no reaction |
| zinc | reaction | reaction | no reaction | reaction | х |

| 1 | Al | uminium is more reactive than zinc and less reactive than magnesium. Would it react with: |
|---|----|--------------------------------------------------------------------------------------------------|
| | a | copper sulfate? |
| | b | magnesium sulfate? |
| 2 | Ti | n is more reactive than copper and less reactive than iron. Would it react with: |
| | a | silver nitrate? |
| | b | zinc sulfate? |
| 3 | Le | ead is more reactive than copper and less reactive than tin. Would it react with: |
| | a | copper sulfate? |
| | b | iron sulfate? |
| 4 | | edium is a very reactive metal. Would a reaction take place between magnesium and sodium loride? |
| 5 | | rite the word equation for the reaction between zinc and copper sulfate. |
| | | |