**Think:**

1. All sodium salts are **…………………………………………….……………………..**
2. All ammonium salts are **……………………………………………………………..**
3. Silver chloride is **……………………………………………………………………….**
4. Potassium iodide is **……………………………………………………………………**
5. All nitrates are **…………………………………………………………………………..**
6. Carbonates are **………………………………………………………………………….**
7. Calcium sulfate is **………………………………………………………………………**
8. **All chlorides are soluble except ………….……………………………………**
9. Ethanoates are **………………………………………………………………………….**
10. Sodium carbonates are **………………………………………………………**

**Complete and balance the equation.**

1. 2Na + 2HCl **………………………………………………………………………[2]**
2. NaOH + HCl **…………………………………………………………………………..**
3. Na2CO3 + 2HCl **……………………………………………………………………[2]**

TOTAL MARKS: \_\_\_/15