

Metal

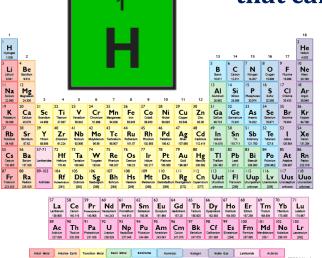


Element



Atom

The smallest particle of a chemical element that can exist









A substance made from only one type of atom. It cannot be broken down by chemical reactions.



THOUGHT BOARD

Look at the **thought board** – what can you say about the pictures?

Can you fit any of the words to the pictures or the definitions?

Are there any pictures you recognise?

Discuss your thoughts with the person next to you

Atom? The smallest particle of a chemical element that can exist

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Element?

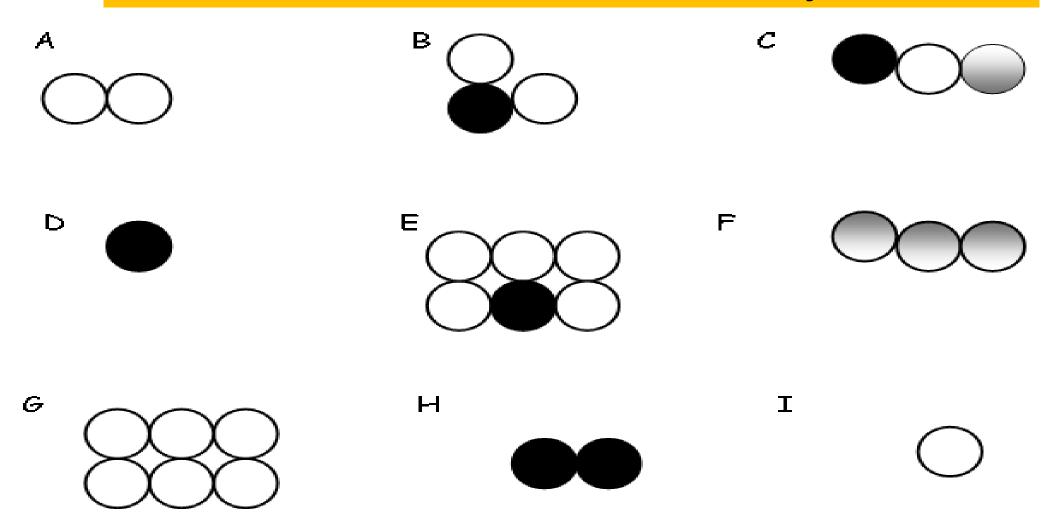
Symbols

Element	Symbol
Carbon	
Hydrogen	
Fluorine	
Calcium	
Nitrogen	
Neon	
Helium	
Sodium	
Potassium	
Chlorine	
Magnesium	

Peer Mark

Recognising elements

Which ones are elements? How do you know?



Describe how the periodic table is organised using the correct terms

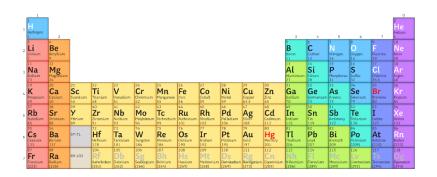
On your blank periodic table, draw the zigzags with a black pen and colour in where the <u>metals</u>, <u>non-metals</u> and <u>metalloids</u> are and label them on your periodic table. Label the <u>alkali metals</u>, <u>transition metals</u>, <u>halogens</u> and <u>noble gases</u>.

1	1 1 Hydrogen 1	2											3	4	5	6	7	2 He Helium
2	3 Li Lithium 7	Be Beryllium											Boron	6 C Carbon 12	7 Nitrogen 14	8 Oxygen 16	9 Fluorine 19	Neon 20
	Na Sodium 23	Mg Magnesium 24											Aluminium 27	Si Silicon 28	Phosphorus	Sulfur	17 Cl Chlorine 35.5	Argon 40
4	19 K Potassium 39	Ca Calcium 40	Sc Scandium 45	Ti Titanium 48	Vanadium	Cr Chromium 52	Mn Manganese 55	Fe Iron 56	Co Cobalt	Ni Nickel 59	Cu	30 Zn Zinc 65	Gallium	Ge Germanium 73	AS Arsenic 75	Se Selenium 79	Br Bromine 80	36 Kr Krypton 84
5	Rb Rubidium 85	38 Sr Stronium 88	39 Yttrium 89	Zr Zirconium	Nb Niobium	MO Molybdenum 96	Tc Technetium	Ruthenium	Rhodium	Palladium	Ag Silver		In In Indium 115		Sb Antimony	Te Tellurium	53 I Iodine 127	Xe Xenon 131
6	Cs	Barium 137	57-71	72 Hf Hafnium 178	Ta Tantalum 181	74 W Tungsten 184	75 Re Rhenium 186	76 Os Osmium	77 Ir Iridium 192	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 TL Thallium 204	Pb Lead		Po Polonium (209)		Rn Radon (222)
7	Fr Francium (223)	Radium (226)	89-103	Rutherfordium (261)	Dubnium (262)	106 Seaborgium (266)	Bh Bohrium (264)	Hassium (269)	109 Mt Meitnerium (268)	DS Darmstadtium (269)	Rg Roentgenium	Cn	Nihomium (286)	114 Flerovium (289)	Mc Mc Moscovium (289)	Livermorium (293)	117 Ts Tennessine (294)	118 Oganesson (294)

Compare elements and suggest similarities of elements in groups within the periodic table



Russian scientist, Mendeleev, put the elements into groups that have similar reactions.



Elements with similar properties are placed in the same group.

For example, **Group 1** contains the alkali metals, which all react quickly with water.

Group 0 elements are all very unreactive.

These chemical properties are called 'characteristics'.

Compare elements and suggest similarities of elements in groups within the periodic table

Where would these elements go on the periodic table?



sodium



sulphur









carbon

How is copper like the others? Where should it go in the table?